

See back cover for an English  
translation of this cover

# 2

L2-CHEMR



NEW ZEALAND QUALIFICATIONS AUTHORITY  
MANA TOHU MĀTAURANGA O AOTEAROA

## Te Mātauranga Matū, Kaupae 2, 2013

9.30 i te ata Rātū 19 Whiringa-ā-rangi 2013

**PUKA RAUEMI**  
mō 91164M, 91165M, 91166M

Tirohia tēnei puka hei whakautu i ngā pātai i roto i ō pukapuka Pātai, Whakautu hoki.

Tirohia mehemea kei roto nei ngā whārangi 2–3 e raupapa tika ana, ā, kāore hoki he whārangi wātea.

**KA TAEA TĒNEI PEPA TE PUPURI HEI TE MUTUNGA O TE WHAKAMĀTAUTAU.**

# TE TAKA PŪMOTU

18

Tau Iraoaho		Papatipu Rāpoi Ngota/g mol <sup>-1</sup>															
1		2															
H 1.0		He 4.0															
3		10															
Li 6.9		Ne 20.2															
4		17															
Be 9.0		F 19.0															
11		18															
Na 23.0		Ar 40.0															
12		36															
Mg 24.3		Kr 83.8															
19		54															
K 39.1		Xe 131															
20		86															
Ca 40.1		Rn 222															
37		210															
Rb 85.5		210															
56		210															
Cs 133		210															
87		210															
Fr 223		210															

57	58	59	60	61	62	63	64	65	66	67	68	69	70
Raupapa Lanthanide	L 139	Ce 140	Pr 141	Nd 144	Pm 147	Sm 150	Gd 157	Tb 159	Tm 169	Dy 163	Ho 165	Er 167	Yb 173
89	90	91	92	93	94	95	96	97	98	99	100	101	102
Raupapa Actinide	Ac 227	Th 232	Pa 231	U 238	Np 237	Pu 239	Cm 244	Bk 249	Cf 251	Es 252	Fm 257	Md 258	No 259

# PERIODIC TABLE OF THE ELEMENTS

Atomic number		Molar mass/g mol <sup>-1</sup>																																	
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18																		
1	2																	3	4																
<b>H</b> 1.0	<b>He</b> 4.0																	<b>Li</b> 6.9	<b>Be</b> 9.0																
																		<b>Na</b> 23.0	<b>Mg</b> 24.3																
																		<b>K</b> 39.1	<b>Ca</b> 40.1																
																		<b>Sc</b> 45.0	<b>Ti</b> 47.9	<b>V</b> 50.9	<b>Cr</b> 52.0	<b>Mn</b> 54.9	<b>Fe</b> 55.9	<b>Co</b> 58.9	<b>Ni</b> 58.7	<b>Cu</b> 63.5	<b>Zn</b> 65.4	<b>Ga</b> 69.7	<b>Ge</b> 72.6	<b>As</b> 74.9	<b>Se</b> 79.0	<b>Br</b> 79.9	<b>Kr</b> 83.8		
																		<b>Y</b> 88.9	<b>Zr</b> 91.2	<b>Nb</b> 92.9	<b>Mo</b> 95.9	<b>Tc</b> 98.9	<b>Ru</b> 101	<b>Rh</b> 103	<b>Pd</b> 106	<b>Ag</b> 108	<b>Cd</b> 112	<b>In</b> 115	<b>Sn</b> 119	<b>Sb</b> 122	<b>Te</b> 128	<b>I</b> 127	<b>Xe</b> 131		
																		<b>Lu</b> 175	<b>Hf</b> 179	<b>Ta</b> 181	<b>W</b> 184	<b>Re</b> 186	<b>Os</b> 190	<b>Ir</b> 192	<b>Pt</b> 195	<b>Au</b> 197	<b>Hg</b> 201	<b>Tl</b> 204	<b>Pb</b> 207	<b>Bi</b> 209	<b>Po</b> 210	<b>At</b> 210	<b>Rn</b> 222		
																		<b>La</b> 139	<b>Ce</b> 140	<b>Pr</b> 141	<b>Nd</b> 144	<b>Pm</b> 147	<b>Sm</b> 150	<b>Eu</b> 152	<b>Gd</b> 157	<b>Tb</b> 159	<b>Dy</b> 163	<b>Ho</b> 165	<b>Er</b> 167	<b>Tm</b> 169	<b>Yb</b> 173				
																		<b>Ac</b> 227	<b>Th</b> 232	<b>Pa</b> 231	<b>U</b> 238	<b>Np</b> 237	<b>Pu</b> 239	<b>Am</b> 241	<b>Cm</b> 244	<b>Bk</b> 249	<b>Cf</b> 251	<b>Es</b> 252	<b>Fm</b> 257	<b>Md</b> 258	<b>No</b> 259				
																		<b>Rb</b> 85.5	<b>Sr</b> 87.6	<b>Y</b> 88.9	<b>Zr</b> 91.2	<b>Nb</b> 92.9	<b>Mo</b> 95.9	<b>Tc</b> 98.9	<b>Ru</b> 101	<b>Rh</b> 103	<b>Pd</b> 106	<b>Ag</b> 108	<b>Cd</b> 112	<b>In</b> 115	<b>Sn</b> 119	<b>Sb</b> 122	<b>Te</b> 128	<b>I</b> 127	<b>Xe</b> 131
																		<b>Fr</b> 223	<b>Ra</b> 226	<b>Ac</b> 227	<b>Th</b> 232	<b>Pa</b> 231	<b>U</b> 238	<b>Np</b> 237	<b>Pu</b> 239	<b>Am</b> 241	<b>Cm</b> 244	<b>Bk</b> 249	<b>Cf</b> 251	<b>Es</b> 252	<b>Fm</b> 257	<b>Md</b> 258	<b>No</b> 259		
																		<b>Cs</b> 133	<b>Ba</b> 137	<b>La</b> 139	<b>Ce</b> 140	<b>Pr</b> 141	<b>Nd</b> 144	<b>Pm</b> 147	<b>Sm</b> 150	<b>Eu</b> 152	<b>Gd</b> 157	<b>Tb</b> 159	<b>Dy</b> 163	<b>Ho</b> 165	<b>Er</b> 167	<b>Tm</b> 169	<b>Yb</b> 173		
																		<b>Fr</b> 223	<b>Ra</b> 226	<b>Ac</b> 227	<b>Th</b> 232	<b>Pa</b> 231	<b>U</b> 238	<b>Np</b> 237	<b>Pu</b> 239	<b>Am</b> 241	<b>Cm</b> 244	<b>Bk</b> 249	<b>Cf</b> 251	<b>Es</b> 252	<b>Fm</b> 257	<b>Md</b> 258	<b>No</b> 259		
																		<b>Li</b> 6.9	<b>Be</b> 9.0																
																		<b>Na</b> 23.0	<b>Mg</b> 24.3																
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																		<b>Sc</b> 45.0	<b>Ti</b> 47.9	<b>V</b> 50.9	<b>Cr</b> 52.0	<b>Mn</b> 54.9	<b>Fe</b> 55.9	<b>Co</b> 58.9	<b>Ni</b> 58.7	<b>Cu</b> 63.5	<b>Zn</b> 65.4	<b>Ga</b> 69.7	<b>Ge</b> 72.6	<b>As</b> 74.9	<b>Se</b> 79.0	<b>Br</b> 79.9	<b>Kr</b> 83.8		
																		<b>Y</b> 88.9	<b>Zr</b> 91.2	<b>Nb</b> 92.9	<b>Mo</b> 95.9	<b>Tc</b> 98.9	<b>Ru</b> 101	<b>Rh</b> 103	<b>Pd</b> 106	<b>Ag</b> 108	<b>Cd</b> 112	<b>In</b> 115	<b>Sn</b> 119	<b>Sb</b> 122	<b>Te</b> 128	<b>I</b> 127	<b>Xe</b> 131		
																		<b>Lu</b> 175	<b>Hf</b> 179	<b>Ta</b> 181	<b>W</b> 184	<b>Re</b> 186	<b>Os</b> 190	<b>Ir</b> 192	<b>Pt</b> 195	<b>Au</b> 197	<b>Hg</b> 201	<b>Tl</b> 204	<b>Pb</b> 207	<b>Bi</b> 209	<b>Po</b> 210	<b>At</b> 210	<b>Rn</b> 222		
																		<b>La</b> 139	<b>Ce</b> 140	<b>Pr</b> 141	<b>Nd</b> 144	<b>Pm</b> 147	<b>Sm</b> 150	<b>Eu</b> 152	<b>Gd</b> 157	<b>Tb</b> 159	<b>Dy</b> 163	<b>Ho</b> 165	<b>Er</b> 167	<b>Tm</b> 169	<b>Yb</b> 173				
																		<b>Ac</b> 227	<b>Th</b> 232	<b>Pa</b> 231	<b>U</b> 238	<b>Np</b> 237	<b>Pu</b> 239	<b>Am</b> 241	<b>Cm</b> 244	<b>Bk</b> 249	<b>Cf</b> 251	<b>Es</b> 252	<b>Fm</b> 257	<b>Md</b> 258	<b>No</b> 259				
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																		<b>Ac</b> 227	<b>Th</b> 232	<b>Pa</b> 231	<b>U</b> 238	<b>Np</b> 237	<b>Pu</b> 239	<b>Am</b> 241	<b>Cm</b> 244	<b>Bk</b> 249	<b>Cf</b> 251	<b>Es</b> 252	<b>Fm</b> 257	<b>Md</b> 258	<b>No</b> 259				

57	58	59	60	61	62	63	64	65	66	67	68	69	70	
Lanthanide Series	<b>La</b> 139	<b>Ce</b> 140	<b>Pr</b> 141	<b>Nd</b> 144	<b>Pm</b> 147	<b>Sm</b> 150	<b>Eu</b> 152	<b>Gd</b> 157	<b>Tb</b> 159	<b>Dy</b> 163	<b>Ho</b> 165	<b>Er</b> 167	<b>Tm</b> 169	<b>Yb</b> 173
Actinide Series	89	90	91	92	93	94	95	96	97	98	99	100	101	102
	<b>Ac</b> 227	<b>Th</b> 232	<b>Pa</b> 231	<b>U</b> 238	<b>Np</b> 237	<b>Pu</b> 239	<b>Am</b> 241	<b>Cm</b> 244	<b>Bk</b> 249	<b>Cf</b> 251	<b>Es</b> 252	<b>Fm</b> 257	<b>Md</b> 258	<b>No</b> 259

*English translation of the wording on the front cover*

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## Level 2 Chemistry, 2013

9.30 am Tuesday 19 November 2013

### RESOURCE SHEET for 91164, 91165, 91166

Refer to this sheet to answer the questions in your Question and Answer Booklets.

Check that this booklet has pages 2–3 in the correct order and that none of these pages is blank.

**YOU MAY KEEP THIS SHEET AT THE END OF THE EXAMINATION.**